



DEPT. OF MECH. IND. & AERO ENG.

#### **MECH 221 - Materials Science**

LECTURES: Wed and Fri H-531 from 10:15 to 11:30 am

#### Instructor: Dr. Mamoun Medraj, P.Eng

e-mail: mmedraj@encs.concordia.ca Dept. of Mech. & Ind. Eng., Room, EV 12.185

Office Hours: Wednesdays 2:30 to 4:00 pm.

#### **TEXTBOOK:**

W.D. Callister, Materials Science & Engineering: An Introduction 5<sup>th</sup> to 9<sup>th</sup> ed., J. Wiley.

Handouts: Available at http://users.encs.concordia.ca/~mmedraj/mech221.html

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# MECH 221 - Materials Science

- Tutorials:
- TA section Friday at 15:15-16:05 pm Rooms: H-429
- TC section Monday at 16:15-17:05 Room: H-562

• TA's:

- Rahele Nikonam
  - email: rahele\_nikonam@ymail.com
- Dulani Kodippili email: dulanipankaja@gmail.com



Assessment:

Exams:

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In-Tutorial Problems 15 %

Midterm

Final

#### What is Material?

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The In-Tutorial Assignment Problems will take place every

second tutorial. The first one will be on Monday Sep. 18th or

Friday Sep. 22nd

The midterm exam is *optional* Students who write the midterm exam, however, will get the higher

mark of the final exam plus the midterm or the final exam alone.

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Midterm Exam: Friday October 20th, 2017

30 %

55 %

- Assignments will not be collected but the questions will be used as the basis for

the *In-Tutorial* Assignment Problems. - Also, some of the assignment problems

and tutorials questions (or similar ones)

will be asked in the exams. The solutions

will be discussed in the tutorial sessions.

Oils, gases, pharmaceuticals : ...... Iron, copper, polymers, cement: .....

- Material can be put into certain geometric shape
- The product has some .....

Old materials: metals, wood, ceramics, skins, natural fibers, (papyrus) New materials: .....

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Mech 221 lecture 1 [2]



# Introduction: Historical Perspective

- Civilization strongly linked with materials Stone age, iron age, bronze age ... nuclear age, information age
  - Sumerians: ceramics
  - Egyptians: lime

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- Anatolians: Iron (12<sup>th</sup> century BC)
- The earliest known Bronze is from what is now Iran and Iraq

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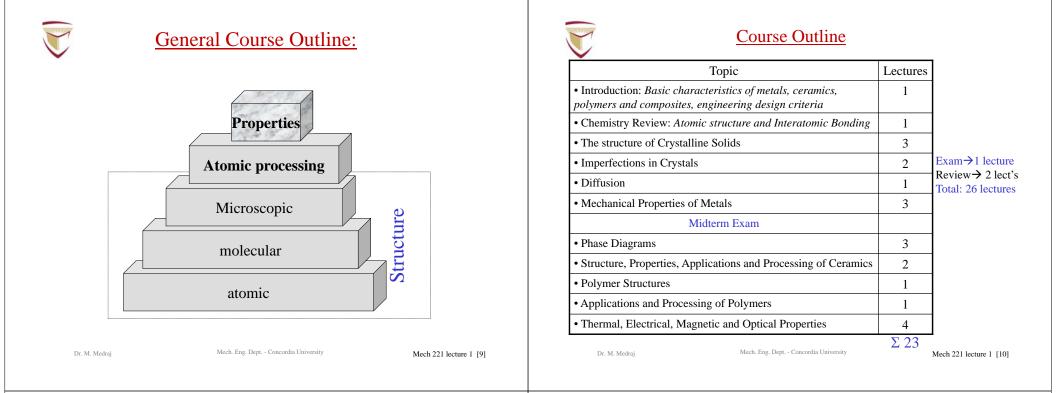


#### Introduction



What is Materials Science ? - Relationships between structure and of materials	Introduction         Property: Response of a material to an external effect, such as
What is Materials Engineering ? - Structure-property correlations - Design the structure of a material to impart some desired properties	<ul> <li>Mechanical</li> <li></li></ul>

Mech 221 lecture 1 [5]





# Why study Materials Science?

(1) Important to understand **capabilities and limitations** of materials:

• The following are just a few examples of catastrophic failure caused by a lack of <u>fundamental understanding</u> of *materials, their properties, and failure modes*.



### Examples of Catastrophic Failure



Liberty ships (WWII)



D-B-T in BCC Fe (metal)



Challenger (1986)

failure of an O-ring seal (polymer)

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# Examples of Catastrophic Failure





Overstressed steel support rods (underdesigned)

Hyatt Regency (KC) walkway collapse (1981)





Excessive wear on stabilizer jackscrew

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# Examples of Catastrophic Failure

• Tacoma Narrows Bridge Collapse (1940)

poor design – .....



- de Havilland Comet (first commercial jet) (1954 55)
   metal fatigue, aggravated by high stresses around rivet holes near window openings
- United DC-10 crash (Sioux City, IA) (1989)

inclusion and cracking in primary #2 engine turbine blade

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# Why Study Materials Science?

(2) An understanding of Materials Science helps us to design better components, parts, devices, etc.

- how do you make something stronger or lighter?
- how do elements come together to form alloys?
- why do some materials have vastly different properties than others?

(3) It is *interesting* and helps to make you a more <u>informed person</u>



# **Classes of Materials**

#### There are 3 major classes:

#### 1. Metals

Pure metallic elements or Combination of metallic elements (alloys) Large number de-localized electrons (conduct electricity)

#### 2. Ceramics

- Molecules based on bonding between metallic and non-metallic elements (including oxides, nitrides, carbides)
- Typically insulating and refractory

#### 3. Polymers

Many are organic compounds Chemically based on C, H, other non-metals Large molecular structures

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