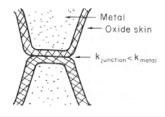


Friction

Complete seizure of metals in vacuum (or reducing atmosphere which removes oxide layer).



Oxide film reduces shear stress required to break junction and reduces μ .

 Material
 µ

 (clean metals in
 Seizure µ >

Material	μ
Perfectly clean metals in	Seizure $\mu > 5$
vacuum	
Clean metals in air	0.8-2
Clean metals in wet air	0.5-1.5
Steel on dry bearing metals	0.1-0.5
(e.g. lead, bronze)	
Steel on ceramics	0.1-0.5
(e.g. sapphire, diamond, ice)	
Ceramics on ceramics	0.05-0.5
(e.g. carbides on carbides)	
Polymers on polymers	0.05-1.0
Metals and ceramics on	0.04-0.5
polymers	
(PE, PTFE, PVC)	
Boundary lubrication of	0.05-0.2
metals	
High-temperature lubricants	0.05-0.2
(MoS ₂ , graphite)	
Hydrodynamic lubrication	0.001-0.005

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Outline

• Review: Friction

• Wear of Materials - Adhesive wear

Wear Prevention

- Abrasive wear - Surface Fatigue

- Corrosion wear

• Design of Journal Bearing

• Materials for Skis and Sledges

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Wear of Materials

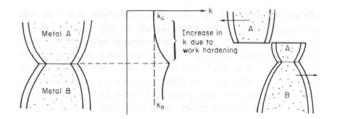
Even when solid surfaces are protected by oxide films and boundary lubricants, some solid-to-solid contact occurs at regions where the oxide film breaks down under mechanical loading, and adsorption of active boundary lubricants is poor. This intimate contact will generally lead to *wear*.

Wear is normally divided into two main types:

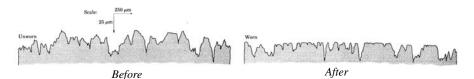
- adhesive wear
- abrasive wear



Adhesive Wear



If the adhesion between A atoms and B atoms is good enough, wear fragments will be removed from the metal A. If materials A and B are the same, wear takes place from surfaces.

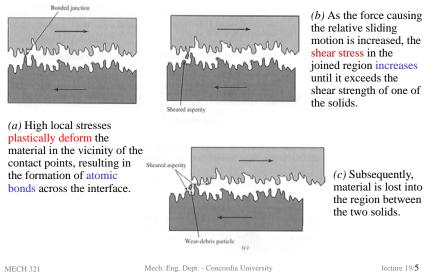


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Adhesive Wear

The sequence of steps occurring during adhesive wear:



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Adhesive Wear

• Metals, with their intermediate values of strength and hardness, are generally resistant to adhesive wear than either:

- the low-strength, high-ductility polymers or

- the high-hardness, low-ductility ceramics.

• The normally incompatible requirements of high hardness and low shear strength make materials ideal candidates for this application. Hence bearing materials are based on this idea:

- E.g. White metal bearings: soft alloys of lead or tin in a matrix of stronger phases;

- bearing bronzes consist of soft lead particles (which smear out to form the lubricating film) supported in a bronze matrix;

- and polymer-impregnated *porous bearings* are made by partly sintering copper with a polymer (usually PTFE) forced into its pores.



Adhesive Wear

• The size of the piece of metal removed from a particular asperity depends on how far away from the junction the shearing takes place.

• If the work-hardened region extends well into the asperity, the tendency will be to produce large pieces;

• This tendency will be reduced if the cross-section of the asperity increases away from the contact region.

• In order to minimise *rate of wear*, we need to minimise the size of metal pieces removed.

1) Minimise the area of contact, a.

Since $a = P/\sigma_v$, reducing the loading on the surfaces will reduce the wear, e.g. chalk on the blackboard

- high load \rightarrow plenty of transfer of chalk to the board- severe wear .

- low load \rightarrow little wear takes place.

2) Increase σ_v , i.e. the *hardness*. - hard pencils write less clearly than soft pencils.

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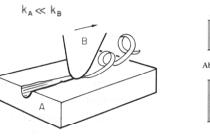
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Abrasive Wear

The loss of surface material is caused by an interaction with a separate particle "trapped" between the two sliding surfaces.



Travel Ø Q ← Debris Abrasive Asperity Worn material

Schematic illustrations of the abrasive wear mechanism.

Abrasive Wear

• It is not unusual for the **abrading particles** to be wear debris from an adhesive mechanism, can also be caused by dirt particles (e.g. sand) making their way into the system, or - in an engine - by combustion products: that is why it is important to filter the oil.

• <u>The rate of material loss</u> is related to the relative hardness of the abrading particles and the sliding surfaces. *If the surface is harder than the particle, the wear rate is minimal.*

• Materials with high hardness, high toughness, and reasonable temperature stability are good candidates for applications requiring high abrasive wear resistance.

• Common selections for this application include tempered martensite, surfacehardened steels, cobalt alloys, and many ceramics.

• Abrasive wear is usually bad - as in machinery - but we would find it difficult to sharpen lathe tools, or polish brass ornaments, or drill rock, without it.

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Corrosive Wear

• The synergistic effects of and degradation can increase the rate of material loss from a solid surface.

• For example, <u>mechanical wear</u> associated with sliding or rolling can break down a protective film and expose the underlying material to an aggressive environment. Alternatively, the <u>corrosion reaction products</u>, particularly hard oxide particles, can serve as abrasive particles.



Corrosive wear



Surface Fatigue

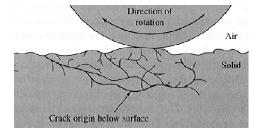
• The previous two wear mechanisms are common during sliding motion.

• If, however, one component is rolling over another, the wear mechanism is often surface fatigue.

• The state of stress developed during rolling contact shows that a shear stress (*Hertz contact stress*) is developed slightly below the surface, and may result in the nucleation of <u>subsurface</u> cracks, which can then propagate to the surface and form wear pits.

• This is a common failure mechanism in railroad wheels.

Wear fatigue mechanism showing the location of the subsurface crack origin



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Wear Prevention

- Common approaches to minimizing wear are:
 - Lubricants
 - Surface-hardening treatments.
 - Wear resistance materials.
- Surface-hardening treatments for reducing wear include:

- *case carburising commonly used in engine crankshafts,*

- ion implantation used in surgical instruments

- hard-faced ceramic coatings used in turbine blades and fibre guides in the textile industry.



Design of Journal Bearings

• In the proper functioning of a well-Lubricated journal bearing, the frictional and wear <u>properties of the</u> <u>materials</u> are, surprisingly, <u>irrelevant</u>.

• The mating surfaces are kept apart by a thin pressurized film of oil formed under conditions of *hydrodynamic lubrication*.

- Working clearance is concentrated on one side.
- Oil is viscous, and is dragged around by shaft.

• The convergence of the oil stream causes an increase in pressure of the oil film, and this pressure literally holds up the shaft against the applied force.

• Pressures of 10 to 100 atmospheres are common under such conditions.

• Journal bearings only operate under hydrodynamic lubrication when the rotational speed of the journal is high enough. When starting an engine up, or running slowly under high load, **hydrodynamic lubrication is**, and, and



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Design of Journal Bearings

• Provided the oil is sufficiently viscous, the film is thick enough to cause complete separation of the mating surfaces:

- there should be no asperity contact, and
- no wear, under ideal hydrodynamic conditions.
- Sliding of the mating surfaces takes place by shear in the liquid oil itself.
- Coefficients of friction (hydrodynamic lubrication) are reduced to 0.001 to 0.005.
- Real bearings contain dirt- (silica), (new automobile engines contain hard cast-iron dust from machining operations).
- If particles are thicker than the oil film at its thinnest abrasive wear will take place.
- Make the mating surfaces harder than the dirt particles. ("case-hardened" to increase the hardness of the surface of the journals/shafts)
- Make the bearing metal soft enough, so dirt particles will be pushed into the surface and out of harm's way. This property of bearing material is called
- Principle of *plastic constraint* is used in bearing design by <u>depositing a very thin layer</u> (about 0.03 mm thick) of soft alloy on to the bearing shell. This is thick enough to embed most dirt particles, but thin enough to support the journal forces.

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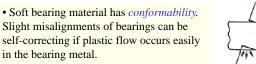
Split shell

construction

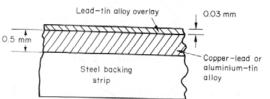


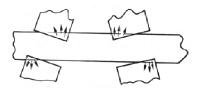
Design of Journal Bearings

- If oil supply fails, frictional heating will rapidly increase the bearing temperature, normally lead to metal-to-metal contact and eventual seizure.
- <u>Soft bearing material</u> (low melting point) will be able to shear and may also melt locally. **Protects** the journal from severe surface damage, and helps to avoid component breakages (*sudden locking of mating surfaces*)



• Clearly there is a compromise between load-bearing ability and conformability.





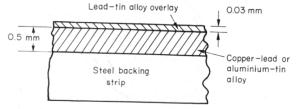


Design of Journal Bearings

• <u>Thin overlay of lead-tin</u> can get worn away under severe operating conditions **before** the end of the normal life of the bearing,

• Customary to put a second thicker, and harder, layer between the overlay and the steel backing strip. The alloys normally used are copper-lead, or aluminium-tin.

• In the event of the wearing through of the overlay they are still soft enough to act as bearing materials <u>without immediate damage</u> to the journal.



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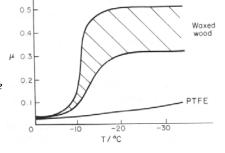


Materials for Skis and Sledge Runners

- Skis, (for people and aircraft), used to be made of waxed wood.
- Down to about ~ -10° C, the friction of waxed wood on snow is very low (is about 0.02).
- Below -10°C, bad things start to happen μ rises sharply to about 0.4.
- Polar explorers have observed this -Wright, (1911-13 Scott expedition), writes:

"Below 0°F (-18°C) the friction (on the sledge runners) seemed to increase progressively as the temperature fell".







Materials for Skis and Sledge Runners

- For ice, its melting point drops if you compress it.
- Many believe that pressure from the skis causes the snow beneath to melt.
- Nonsense: large man lowers the melting point by only 0.1°C at asperities: (10⁻³ of nominal area),

• Work is done against the frictional forces; heat is generated at the sliding surface, sufficient to melt a layer of ice, **producing a thin film of water**, at points where asperities touch the ski.

• The skier hydroplanes along on a layer of water generated by his/her own friction.

1 atm solid und vapour und vapour

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lecture 19/18



Materials for Skis and Sledge Runners

Why skis with exposed metal (aluminium or steel edges) are slower at low temperatures than those without?

- Below -10°C, heat is conducted away too quickly to allow melting
- The mechanism of friction is the same as that of metals: where ice asperities adhere to the ski and must be sheared when it slides.
- Coefficient of friction is large (0.4) (*ski-planes, skis, sledges etc have problems*)
- When ceramics (ice is a ceramic) slide on polymers, μ can be as low as 0.04.
- PTFE ("Teflon ") and polyethylene have very low μ values.
- By coating the ski or sledge runners with these materials, the coefficient of friction stays low, *even* when the temperature is so low that frictional heating is unable to produce a boundary layer of water.

• Aircraft and sports skis now have polyethylene or Teflon under surfaces.





Next time: Materials Selection