



Outline

Examples on:

- Fracture Mechanics
- Fatigue
- Wear rate
- Electrochemical cell
- Materials selection



Example 1: *Fracture Mechanics*

A sharp penny-shaped crack with a diameter of 2.5-cm is completely embedded in a solid. Catastrophic fracture occurs when a stress of 700 MPa is applied.

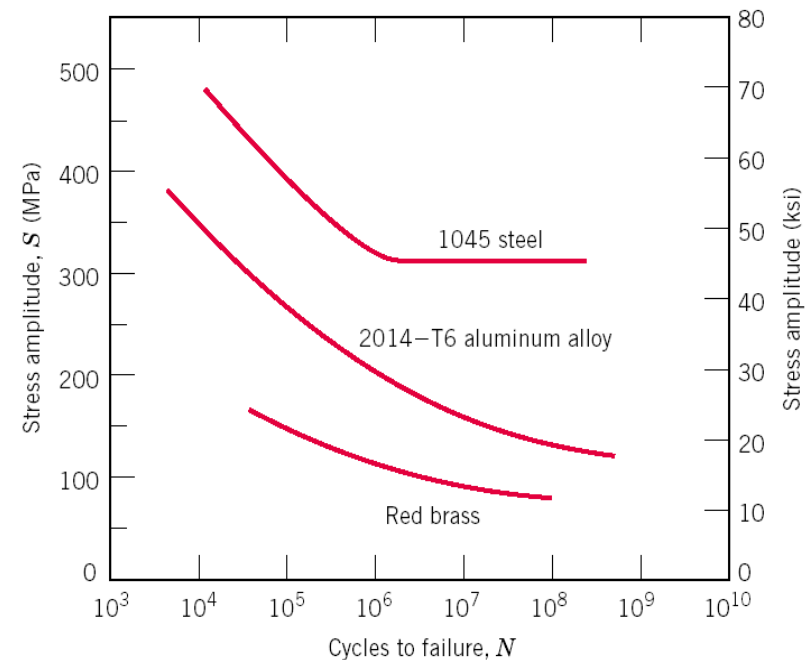
(a) What is the fracture toughness for the material? (*Assume that this value is for plane-strain conditions. The value of Y in this configuration is $2/\pi$*).

(b) If a sheet (0.75 cm thick) of this same material is prepared for fracture-toughness testing (width $B = 0.75$ cm), would the fracture-toughness value measured be a valid number for plane strain fracture-toughness (K_{IC}) or not? If not, what would be the minimum thickness for valid K_{IC} determination? *The yield strength of the material is 1100 MPa.*



Example 2: Fatigue

Consider the 2014-T6 aluminum alloy the S-N behaviour for which is shown in the accompanying figure. A 0.505 in. diameter cylindrical rod fabricated from this alloy is subjected to a repeated tension-compression loading along its axis. Determine the fatigue life if the maximum and mean loads are 8,000 lb_f and 2,000 lb_f, respectively. Assume that the stress plotted on the vertical axis is stress amplitude and that the curve was generated for a mean stress corresponding to a mean load of 2,000 lb_f.





Example 3: *Wear*

Wear in metals is sometimes rapid at first but then appears to lessen as the time in service increases. How this non-uniform wear rate might occur?



Example 4: Electrochemical Cell

An electrochemical cell is composed of pure copper and pure lead electrodes immersed in solutions of their respective divalent ions. For a $6 \times 10^{-3} M$ concentration, of Cu^{2+} , the lead electrode is oxidized yielding a cell potential of 0.4 V. Calculate the concentration of Pb^{2+} ions if the temperature is 50°C .

Table 18.1 The Standard emf Series *5th ed.*

	<i>Electrode Reaction</i>	<i>Standard Electrode Potential, V^0 (V)</i>
	$\text{Au}^{3+} + 3e^- \longrightarrow \text{Au}$	+1.420
	$\text{O}_2 + 4\text{H}^+ + 4e^- \longrightarrow 2\text{H}_2\text{O}$	+1.229
	$\text{Pt}^{2+} + 2e^- \longrightarrow \text{Pt}$	~+1.2
	$\text{Ag}^+ + e^- \longrightarrow \text{Ag}$	+0.800
	$\text{Fe}^{3+} + e^- \longrightarrow \text{Fe}^{2+}$	+0.771
Increasingly inert (cathodic) ↑	$\text{O}_2 + 2\text{H}_2\text{O} + 4e^- \longrightarrow 4(\text{OH}^-)$	+0.401
	$\text{Cu}^{2+} + 2e^- \longrightarrow \text{Cu}$	+0.340
	$2\text{H}^+ + 2e^- \longrightarrow \text{H}_2$	0.000
	$\text{Pb}^{2+} + 2e^- \longrightarrow \text{Pb}$	-0.126
	$\text{Sn}^{2+} + 2e^- \longrightarrow \text{Sn}$	-0.136
	$\text{Ni}^{2+} + 2e^- \longrightarrow \text{Ni}$	-0.250
	$\text{Co}^{2+} + 2e^- \longrightarrow \text{Co}$	-0.277
	$\text{Cd}^{2+} + 2e^- \longrightarrow \text{Cd}$	-0.403
	$\text{Fe}^{2+} + 2e^- \longrightarrow \text{Fe}$	-0.440
Increasingly active (anodic) ↓	$\text{Cr}^{3+} + 3e^- \longrightarrow \text{Cr}$	-0.744
	$\text{Zn}^{2+} + 2e^- \longrightarrow \text{Zn}$	-0.763
	$\text{Al}^{3+} + 3e^- \longrightarrow \text{Al}$	-1.662
	$\text{Mg}^{2+} + 2e^- \longrightarrow \text{Mg}$	-2.363
	$\text{Na}^+ + e^- \longrightarrow \text{Na}$	-2.714
	$\text{K}^+ + e^- \longrightarrow \text{K}$	-2.924

17.1 in the 6th ed.



Example 5: Materials Selection

Select from this list the one metal or alloy that is best suited for each of the following applications, and cite at least one reason for your choice:

- (a) The base for a milling machine.
- (b) The walls of a steam boiler.
- (c) High-speed aircraft.
- (d) Drill bit.
- (e) Cryogenic (i.e., very low temperature) container.
- (f) As a pyrotechnic (i.e., in flares and fireworks).
- (g) High-temperature furnace heating elements to be used in oxidizing atmospheres.

Below is a list of metals and alloys:

- *Plain carbon steel*
- *Magnesium*
- *Brass*
- *Zinc*
- *Tungsten*
- *Gray cast iron*
- *Tool steel*
- *Platinum*
- *Aluminum*
- *Titanium alloy*
- *Stainless steel*



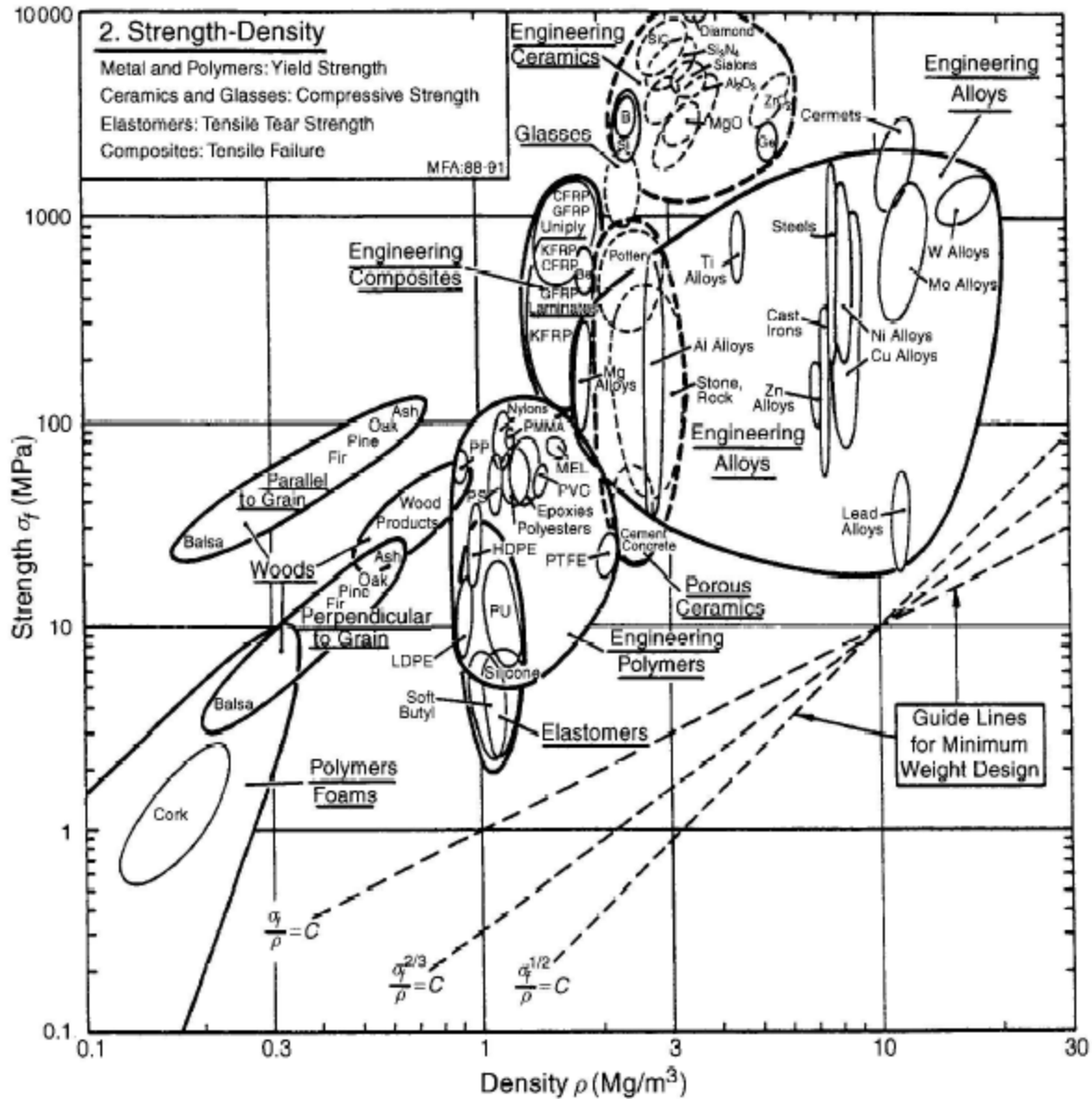
Example 6: *Materials Selection*

The connecting rod of an internal combustion engine is essentially a column loaded in compression.

- a) Ignoring elevated temperature concerns, derive the material indices to minimize mass while simultaneously avoiding plastic yielding and buckling.
- b) Use the materials indices derived in (a) in conjunction with the appropriate materials selection chart to propose the best materials for the connecting rod.
- c) List (at least) two material properties that do not enter into the above analysis but that would be important to the overall design.
- d) If a limit is imposed such that the fracture toughness of the material must be greater than $10 \text{ Mpa}\cdot\text{m}^{-0.5}$ how does this change your answer?

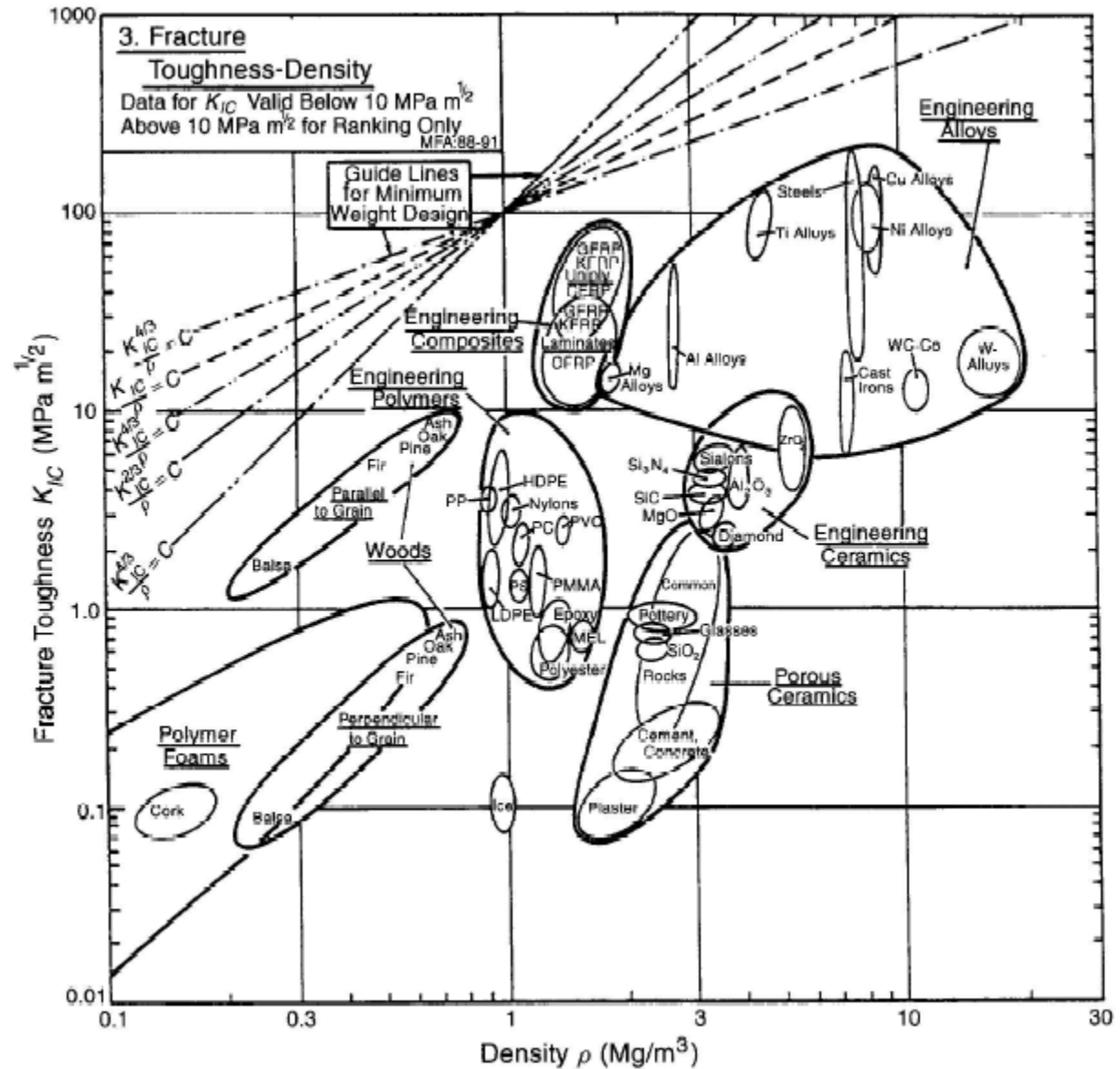


Example 6: *Materials Selection*





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Thanks for listening
Good luck in your exams