

Cu

 $cF4$  $(225) Fm-3m - a$ 

**Cu** [1], c.c.p. (cubic close-packed), f.c.c. (face-centered cubic), Strukturbericht notation A1

Structural features: Close-packed layers in c stacking, i.e. consecutive layers are shifted in the same direction. See Fig. I.37.

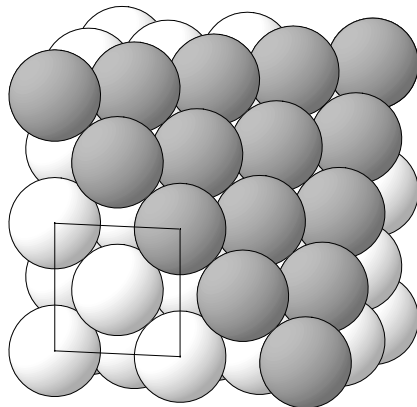


Fig. I.37. **Cu**

Arrangement of Cu atoms emphasizing a close-packed layer perpendicular to [111].

Bragg W.L. (1914) [1]

Cu

$a = 0.36 \text{ nm}$ ,  $V = 0.0467 \text{ nm}^3$ ,  $Z = 4$

site	Wyck.	sym.	$x$	$y$	$z$	occ.	atomic environment
Cu1	$4a$	$m-3m$	0	0	0		cuboctahedron $\text{Cu}_{12}$

Experimental: single crystal, spectrometer, X-rays

References: [1] Bragg W.L. (1914), Philos. Mag. 28, 355-360.

**MnCu<sub>2</sub>Al** [2], Heusler phase, Strukturbericht notation L2<sub>1</sub>; NaLi<sub>2</sub>Sb [3]; LiRuGa<sub>2</sub> [5]; CsLi<sub>2</sub>C<sub>60</sub> [4], fulleride-CsLi<sub>2</sub>

Structural features: Cu(Mn<sub>4</sub>Al<sub>4</sub>) cubes share all faces to form a 3D-framework. Alternatively: AlCu<sub>8</sub> and MnAl<sub>8</sub> cubes share all faces to form a 3D-framework. Ordering variant of W (b.c.c.), sequence along <111>: -Cu-Al-Cu-Mn-. NaLi<sub>2</sub>Sb contains close-packed Sb layers in c stacking, Na occupies all octahedral and Li all tetrahedral voids. See Fig. I.41.

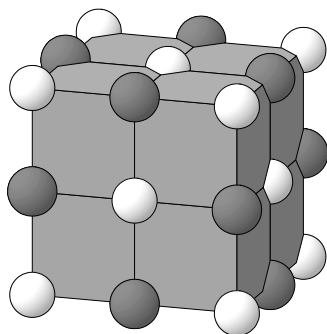


Fig. I.41. **MnCu<sub>2</sub>Al**

Arrangement of Cu(Mn<sub>4</sub>Al<sub>4</sub>) cubes (Mn atoms dark, Al atoms light).

Soltys J. (1981) [1]

AlCu<sub>2</sub>Mn

$a = 0.5958 \text{ nm}$ ,  $V = 0.2115 \text{ nm}^3$ ,  $Z = 4$

site	Wyck.	sym.	$x$	$y$	$z$	occ.	atomic environment
Cu1	8c	-43m	1/4	1/4	1/4		rhombic dodecahedron Al <sub>4</sub> Mn <sub>4</sub> Cu <sub>6</sub>
Mn2	4b	<i>m-3m</i>	1/2	1/2	1/2		rhombic dodecahedron Cu <sub>8</sub> Al <sub>6</sub>
Al3	4a	<i>m-3m</i>	0	0	0		rhombic dodecahedron Cu <sub>8</sub> Mn <sub>6</sub>

Experimental: powder, diffractometer, X-rays

Remarks: The centers of pseudo-spherical C<sub>60</sub> units are considered for CsLi<sub>2</sub>C<sub>60</sub> (orientational disorder). In the abstract of [3] the Wyckoff positions of the Li and Na sites are misprinted as 8*b* and 4*a* instead of 8*c* and 4*b*, respectively.

References: [1] Soltys J. (1981), Phys. Status Solidi A 66, 485-491. [2] (1931), Strukturberichte 1, 488, 551. [3] Hurng W. (1991), Mater. Res. Bull. 26, 439-442. [4] Hirotsawa I., Prassides K., Mizuki J., Tanigaki K., Gevaert M., Lappas A., Cockcroft J.K. (1994), Science (Washington D.C.) 264, 1294-1297. [5] Czybulka A., Petersen A., Schuster H.U. (1990), J. Less-Common Met. 161, 303-312.

**Th<sub>6</sub>Mn<sub>23</sub>** [1], Strukturbericht notation D8<sub>a</sub>

Structural features: Intergrowth of Ca<sub>3</sub>Ag<sub>8</sub>- (Th<sub>6</sub> octahedron surrounded by a Mn<sub>8</sub> cube and a Mn<sub>12</sub> cuboctahedron) and W- (a central Mn atom surrounded by a Mn<sub>8</sub> cube, a Th<sub>6</sub> octahedron and a Mn<sub>12</sub> cuboctahedron) type nested polyhedra units. See Fig. I.53.

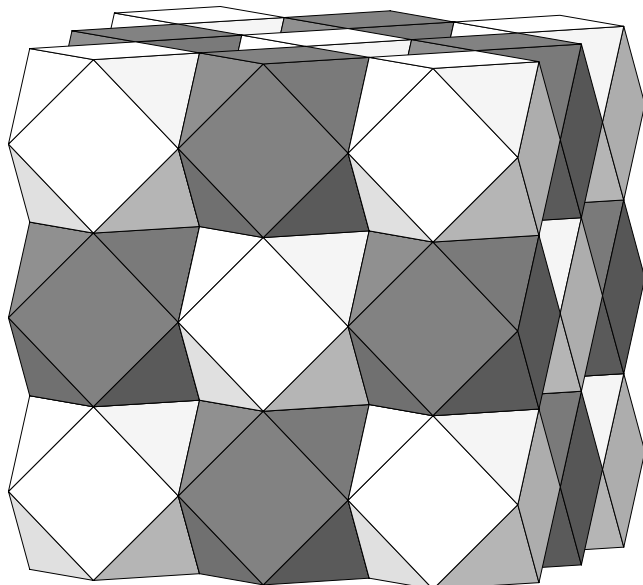


Fig. I.53. **Th<sub>6</sub>Mn<sub>23</sub>**

Arrangement of Mn<sub>12</sub> cuboctahedra, which represent 27-atom W-type (Mn central atom + Mn<sub>8</sub> cube + Th<sub>6</sub> octahedron + Mn<sub>12</sub> cuboctahedron; light) and 26-atom Ca<sub>3</sub>Ag<sub>8</sub>-type (Th<sub>6</sub> octahedron + Mn<sub>8</sub> cube + Mn<sub>12</sub> cuboctahedron; dark) nested polyhedra units.

Florio J.V. et al. (1952) [1]

Mn<sub>23</sub>Th<sub>6</sub>

$a = 1.2523 \text{ nm}$ ,  $V = 1.9639 \text{ nm}^3$ ,  $Z = 4$

site	Wyck.	sym.	$x$	$y$	$z$	occ.	atomic environment
Mn1	32 <i>f</i>	.3 <i>m</i>	0.122	0.122	0.122		pseudo Frank-Kasper Mn <sub>10</sub> Th <sub>3</sub>
Mn2	32 <i>f</i>	.3 <i>m</i>	0.322	0.322	0.322		icosahedron Mn <sub>9</sub> Th <sub>3</sub>
Th3	24 <i>e</i>	4 <i>m.m</i>	0.297	0	0		7-capped pentagonal prism Mn <sub>13</sub> Th <sub>4</sub>
Mn4	24 <i>d</i>	<i>m.mm</i>	0	$\frac{1}{4}$	$\frac{1}{4}$		icosahedron Mn <sub>8</sub> Th <sub>4</sub>
Mn5	4 <i>a</i>	<i>m-3m</i>	0	0	0		square prism (cube) Mn <sub>8</sub>

Transformation from published data: origin shift  $\frac{1}{2} \frac{1}{2} \frac{1}{2}$

Experimental: single crystal, Weissenberg and rotation photographs, X-rays, R = 0.100

References: [1] Florio J.V., Rundle R.E., Snow A.I. (1952), Acta Crystallogr. 5, 449-457.